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Phytochemical Analysis of *Jasminum auriculatum* Vahl. Using FT-IR and GC-MS

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Abstract

The aim at the present study was to investigate the ethanol extract from *Jasminum auriculatum* Vahl. Qualitative analysis showed the presence of alkaloids, carbohydrates, flavonoids, tannins, phenol, steroids, terpenoids and glycosides in the different leaf extracts. The leaf extract did not contain proteins, mucilage, saponins, fats and fixed oil. This work deals with the phytochemical screening and GC-MS studies of the ethanol extract. The highest peak area (35.16%) was obtained by n-Hexadecanoic acid (C₁₆H₃₂O₂) at retention time of 17.275 and the lowest peak area (0.23%) was obtained by Octacosane (C₂₈H₅₈) at retention time of 30.441. The FT-IR spectroscopy studies shows different characteristic peak values with many functional groups present like alkyl halides, aromatics, aliphatic amines, alkanes, amines, unsaturated esters, anhydride, alkene, alkynes, nitriles, carboxylic acids alcohol and phenol. The results important bioactive compounds leaf extracts and these constituents may be responsible for pharmacological activities.

Keywords: *Jasminum auriculatum*, Preliminary Phytochemical, GC-MS and FT-IR analysis.

Introduction

Medicinal plants have been used by human beings since ages in traditional medicine due to their therapeutic potential. The research on medicinal plants has led to the discovery of novel drug used against diverse diseases. According to the World Health Organization's report that nearly 65-80% of world's population in developing countries depends on the traditional medicine for their primary health care and treatment of ailments. The reasons for wide acceptance of herbal medicines are of their being comparatively less expensive, lesser side effects and being natural in origin and hence socially and culturally acceptable (Muniappan Ayyanar *et al.*, 2011) [21]. Higher plants are sources of bioactive compounds continue to play a dominant role in the maintenance of human health. Reports available on the green plants to represent a reservoir of effective chemotherapeutics, which are non-phytotoxic, more systemic and easily biodegradable (Kaushik *et al.*, 2002; Chaman Lal *et al.*, 2006) [16, 7]. Plants are rich source of secondary metabolites with interesting biological activities. In general these secondary metabolites are important source with a variety of structural arrangements and properties (De-Fatima *et al.*, 2006) [9].

Jasminum auriculatum is a shrub used in traditional medicines, Ayurveda, Siddha and Unani (Vaidyaratnam, 2003; Bedi, 2008) [26, 6]. Studies conducted on this species showed that it possessed beneficial effects as suppurative, skin diseases, thermogenic, urolithiasis, ulcers and wounds. The present review highlights the various folk, ayurvedic uses, pharmacognostical, phytochemical and pharmacological studies conducted on *Jasminum auriculatum* and also the unexplored potential of the plant (Rajinder Raina *et al.*, 2008) [23].

The aqueous and alcoholic extracts from flowers of *Jasminum auriculatum* were found to be effective against ethylene glycol induced lithiasis by reducing and preventing the growth of urinary stone (Bahuguna *et al.*, 2009 a) [5]. It also exhibits diuretic effects at dose of 250 mg body weight by increasing the total volume of urine and concentrations of potassium and sodium salts in urine as compared to Frusemide. The alcoholic extract showed more effect (94.81%) (Bahuguna *et al.*, 2009 b) [4] Then the aqueous extract (91.81%). More precise information about qualitative analysis can be obtained by gas-chromatography coupled with mass spectrometry (GC-MS) (Cong *et al.*, 2007) [8]. For quantitative determination, gas chromatography with flame ionization detectors (GC-FID) and GC-MS are preferred (Lee *et al.*, 2005; Lampronti *et al.*, 2006; Haznagy Radnal *et al.*, 2007) [19, 18, 15].

Traditional medicine is an important source of potentially useful compounds for the development of chemotherapeutic agents (Palombo *et al.*, 2001) [22]. A wide range of medicinal

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plant parts is used for extracting from raw drugs and they possess various medicinal properties (Uniyal *et al.*, 2006) [25]. In developing countries the practice of medicine still relies heavily on plant and herbal extract for the treatment of human ailments. Dietary agents consists of a wide variety of biologically active compounds that are ubiquitous in plants and many of them have been used as traditional medicines (Aggarwal *et al.*, 2003; Aggarwal *et al.*, 2004; Aggarwal *et al.*, 2006) [1, 2, 3]. The root is useful in skin diseases especially for ring-worm. The flowers are useful in burning sensation, hyperdesia, ulcers, odontalgia, stomatopathy, ophthalmopathy, cardiopathy, urolithiasis, nephrolithiasis, strangury and dermatopathy (Ghosh, 1984) [12]. *Jasminum auriculatum* leaves have been reported to contain lupeol and jasminol (Deshpande *et al.*, 1967) [10]. The present work to investigate the phytochemicals present in the different leaf extract qualitatively by applying phytochemical test and quantitatively by Gas Chromatography- Mass Spectroscopy (GC-MS) and Fourier Transform Infrared Spectroscopic Analysis (FT-IR) analysis.

2. Materials and Methods

2.1 Collection and preparation of plant materials

The fresh leaf of *Jasminum auriculatum* Vahl was collected from the natural habitats of Villupuram District, Tamilnadu, India. The specimen *Jasminum auriculatum* Vahl, was identified authentication using the specimen- BWH 78213 housed in the Botany wing, Annamalai University, Tamilnadu. The leaf was washed thoroughly 3 times with running tapwater to remove soil particles and adhered debris and finally with sterile distilled water. The leaf was cut and dried in the shadow ground into fine powder and stored in the air tight polythene bags for use.

2.2 Extraction and processing

The shade dried plant material (200g) was crushed in Soxhlet extractor using petroleum ether, chloroform, ethanol, methanol successively (Green, 2004) [13]. The extracts were filtered using Whatman 41 filter paper and were concentrated. The concentrated methanol and ethanol extracts were subjected to qualitative phytochemical analysis and GC-MS analysis.

2.3 Chemical reagents

Chemicals and reagents like petroleum ether, chloroform, ethyl acetate, ethanol, sulphuric acid, hydrochloric acid, Mayer's reagents, Ninhydrin reagents, Fehling solutions A and B, Ferric chloride solution, gelatin, lead acetate, sodium hydroxide, were used to analyze phytochemicals present in the aerial parts of *Jasminum auriculatum* Vahl. Chemical purchased from Sd FiNe-Chem Limited, Mumbai.

3. Screening of phytochemical constituents

Phytochemical screening for the extracts was carried out as per standard methods prescribed by Harbone (1973) [14], Treas and Evans (1989) [24] to find out the presence of various constituents in different leaf extracts.

4. GC-MS Analysis

4.1 Preparation of extract:

About 25 g of the powdered leaf was soaked in 95% methanol for 12h. The extracts were then filtered through Whatman 41 filter paper along with 2g sodium sulphate to remove the sediments and traces of water in the filtrate. Before filtering, the filter paper was made wet with 95% ethanol along with sodium sulphate. The filtrate was then concentrated on

bubbling nitrogen gas into the solution. The extract contained both polar and non-polar phyto-components in the plant material. 2 μ l of this solution was employed for GC-MS analysis (Merlin *et al.*, 2009) [20]. GC-MS analysis of these extracts was performed using a Perkin-Elmer GC Clarus 500 system and Gas chromatograph interfaced to a Mass spectrometer (GC-MS) equipped with an Elite-I, fused silica capillary columns (30mmX0.25mm 1D X 1 μ Mdf, composed of 100% Dimethyl poly siloxane). For GC-MS detection, an electron ionization system with ionizing energy of 70 eV was used. Helium gas (99.999%) was used as the carrier gas at constant flow rate of 1ml/min and an injection volume of 2 μ l was employed (split ratio of 10:1) Injector temperature 250 $^{\circ}$ C Ionsource temperature 280 $^{\circ}$ C. The oven temperature was programmed from 110 $^{\circ}$ C (isothermal for 2 min.), with an increase of 10 $^{\circ}$ C/min to 200 $^{\circ}$ C, then 5 $^{\circ}$ C/min to 280 $^{\circ}$ C, ending with a 9min isothermal at 280 $^{\circ}$ C. Mass spectrum was taken at 70 eV a scan interval of 0.5sec and fragment from 45 to 450 Da. Total GC running time was 36 min. The relative % amount of each component was calculated by comparing its average peak area to the total areas, software adopted to handle mass spectra and chromatograms was a Turbomass.

4.2 Identification of components:

Identification was based on the molecular structure, molecular mass and calculated fragments. Interpretation of mass spectrum GC-MS was conducted using the database of National Institute Standard and Technology (NIST) having more than 62,000 patterns. The name, molecular weight and structure of the components of the test materials were ascertained. The relative percentage amount of each component was calculated by comparing its average peak area to the total areas. The spectrum of the unknown component was compared with the spectrum of the component stored in the NIST library version (2005), software, Turbomas 5.2. This is done in order to determine whether this plant species contains any individual compound or group of compounds, which may substantiate its current commercial and traditional use as herbal medicine. Further it helps to determine the most appropriate methods of extracting these compounds. These results will consequently be discussed in the light of their putative biological or therapeutic relevance.

5. Fourier Transform Infrared Spectroscopic Analysis (FT-IR)

The fresh leaf of *Jasminum auriculatum* was oven dried at 60 $^{\circ}$ C and ground into fine powder using a mortar and pestle. Two milligrams of the sample was mixed with 100mg KBr (FT-IR grade) and then compressed to prepare a salt disc (3mm diameter). The disc was immediately kept in the sample holder and FT-IR spectra were recorded in the absorption range between 4000 and 400 cm^{-1} . All investigations were carried out with a Shimadzu FT-IR spectrometer.

6. Results and Discussion

The phytochemical screening for crude extracts of from leaf *Jasminum auriculatum*. Table.1 represents the phytochemical *Jasminum auriculatum* Vahl and showed that the presence or absent of alkaloids, saponins, tannins, terpenoids, glycosides, carbohydrates, flavonoids, phenol and steroids.

The FT-IR spectrum was used to identify and detect the characteristic peaks and functional groups of the active components based on the peak value in the region of infrared radiation (Table 2; Figure 1). The results revealed the presence of different phytochemicals which are formed during the plants normal metabolic processes. The extract of

Jasminum auriculatum was subjected to FT-IR analysis and the functional groups of the components were separated based on their peak ratios. The result confirmed the presence of (C–Br stretch) normal alkyl halides, (C–H) aromatics, (C–N stretch) aliphatic amines, (C–N stretch) aromatic amines, (C–H rock) alkanes, (C–C stretch) aromatics, (N–H bend) 1° amines, (C=O stretch) α,β -unsaturated esters, (C=C stretch) anhydride (C=C stretch) alkene, (C≡C stretch) alkynes, (C≡C

stretch) nitriles, (O–H stretch) carboxylic acids, (O–H stretch, H–bonded) alcohol and phenols. Which showed major peaks at 524.64, 775.38, 893.04, 1037.7, 1153.43, 1253.73, 1321.24, 1386.82, 1409.96, 1627.92, 1730.15, 1878.67, 1915.31, 2166.06, 2200.78, 2364.73, 2856.58, 2924.09, 3363.86, 3657.04, 3809.41, 3824.84, 3846.06, 3880.78, 3919.35 and 3959.86, cm^{-1} respectively (Figure 2; Table 4).

Table 1: Preliminary Phytochemical screening of *Jasminum auriculatum* Vahl

| S.No | Chemical constituents | Test | Leaf extract | | | |
|------|-----------------------|------------------------|-----------------|------------|---------------|---------|
| | | | Petroleum ether | Chloroform | Ethyl acetate | Ethanol |
| 1 | Alkaloids | Mayer's Test | + | - | + | + |
| 2 | Carbohydrates | Fehling's Test | - | + | - | + |
| 3 | Flavanoids | Chinasoda Test | - | + | - | + |
| 4 | Tannins | FeCl ₃ test | - | - | + | + |
| 5 | Phenol | Gelatin test | - | - | + | + |
| 6 | Protein | Ninhydrin test | - | - | - | - |
| 7 | Gum/Mucilage | Molish Test | - | - | - | - |
| 8 | Steroids | Sulphuric acid Test | - | + | + | + |
| 9 | Terpenoids | Salkowaski test | - | - | + | + |
| 10 | Glycoside | Borntrager Test | - | - | - | + |
| 11 | Saponins | Foam Test | - | - | + | + |
| 12 | Fats and fixed oils | Spot Test | - | - | - | - |

GC-MS analysis of the extracts

During the GC-MS analysis of *Jasminum auriculatum* revealed the presence of 14 compounds identified in the ethanolic extract. The GC-MS identification of phytochemical compounds is based on the peak area molecular weight and molecular formula. The chromatogram (Figure.1) of ethanol leaf extract shows 6 prominent peaks as ethanol extract shows 6 prominent peaks as n-Hexadecanoic acid (C₁₆H₃₂O₂) with retention times of 17.275 and peak area of 35.16 and Cis cis-7,10,13- Hexadecatrienal (C₁₆H₂₆O)with retention time of 18.985 and peak area 33.50 and Phytol (C₂₀H₄₀O)with

retention time of 18.608 and peak area of 12.85 and 2,6,10,14,18,22-Tetracosahexaene,2,6,10,19,23, hexamethyl (C₃₀H₅₀)with retention time 27.060 and peak area of 5.28 and Di-n-octylphthalate (C₂₄H₃₈O₄) with retention time of 23.076 and peak area 4.75and 8, 11,14,Eicosatrienoic acid (Z,Z,Z) (C₂₀H₃₄O₂)with retention time of 19.099 and peak area 2.10. The other less prominent peaks at other retention times are shown in Table 2. The total ions chromatograph (TIC) showing the peak identities of the various compounds are identified as shown in Figure 1. The structure of compounds of ethanol leaf extract are presented in Table 3.

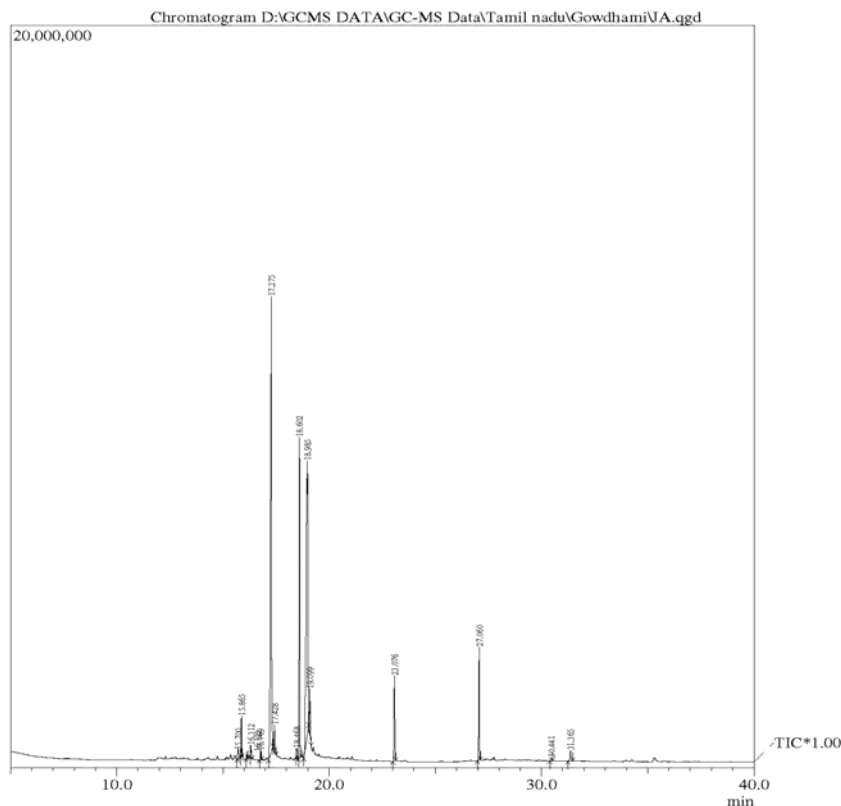


Fig 1: GC-MS analysis of *Jasminum auriculatum* leaves of ethanolic extract

Table 2: Chemical constituents present in the ethanolic extracts of *Jasminum auriculatum* Vahl.

| Peak number | Retention Time | Name of the compound | Molecular formula | Molecular weight | Peak area % |
|-----------------------------|----------------|--|--|------------------|-------------|
| Ethanol leaf extract | | | | | |
| 1 | 15.700 | R-Limonene | C ₁₀ H ₁₆ O ₃ | 184 | 0.78 |
| 2 | 15.865 | 1-Octadecyne | C ₁₈ H ₃₄ | 250 | 1.53 |
| 3 | 16.312 | 1-Octadecyne | C ₁₈ H ₃₄ | 250 | 0.44 |
| 4 | 16.769 | Eicosanoic acid, Methyl ester | C ₂₁ H ₄₂ O ₂ | 326 | 0.39 |
| 5 | 17.275 | n-Hexadecanoic acid | C ₁₆ H ₃₂ O ₂ | 256 | 35.16 |
| 6 | 17.428 | Ethyl nonadecanoate | C ₂₁ H ₄₂ O ₂ | 326 | 1.18 |
| 7 | 18.468 | Methyl 9-Octadecenoate | C ₁₉ H ₃₆ O ₂ | 296 | 0.64 |
| 8 | 18.602 | Phytol | C ₂₀ H ₄₀ | 296 | 12.85 |
| 9 | 18.985 | Cis,cis,cis-7,10,13-Hexadecatrienal | C ₁₆ H ₂₆ O | 234 | 33.50 |
| 10 | 19.099 | 8,11,14-Eicosatrienoic acid,(Z,Z,Z)- | C ₂₀ H ₃₄ O ₂ | 306 | 2.10 |
| 11 | 23.076 | Di-n-octyl phthalate | C ₂₄ H ₃₈ O ₄ | 390 | 4.75 |
| 12 | 27.060 | 2,6,10,14,18,22,-Tetracosahexaene,2,6,10,15,19,23-hexamethyl | C ₃₀ H ₅₀ | 410 | 5.28 |
| 13 | 30.441 | Octacosane | C ₂₈ H ₅₈ | 394 | 0.23 |
| 14 | 31.365 | 2,5,7,8-Tetramethyl-2-(4,8,12-Trimethyltridecyl | C ₂₉ H ₅₀ O ₂ | 430 | 0.89 |

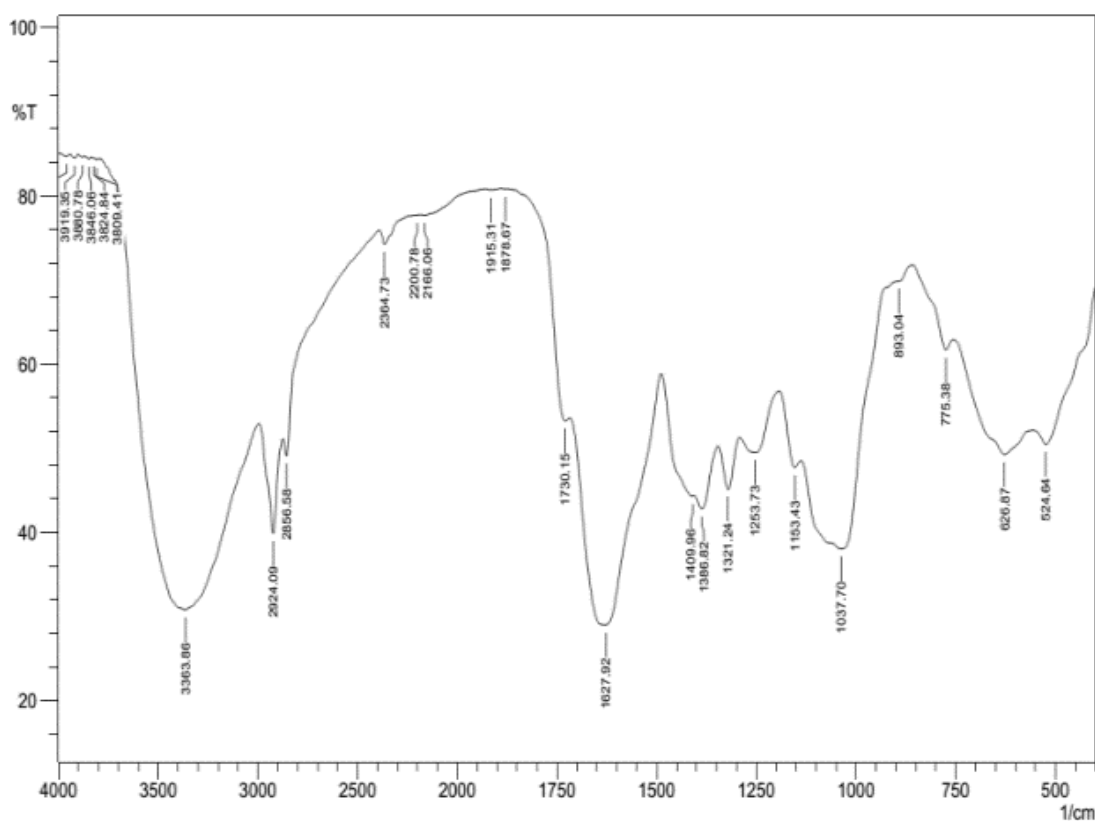


Fig 2: FT-IR Spectrum analysis of *Jasminum auriculatum* leaves of ethanolic extract

This study highlights the presence of many secondary metabolites in the leaves of *Jasminum auriculatum* Vahl, provide an overview of the different classes of molecules present that have led to their pharmacological activities. This study confirmed that the plant extract could be used for the treatment of various diseases. The GC-MS analysis of leaves extract showed the presence of various types of compounds in *Jasminum auriculatum*. Compounds like Phytol (C₂₀H₄₀O) having anti-cancerous activity (Yuenyongsawad *et al.*,2005) [28], n-Hexadecanoic acid (C₁₆H₃₂O₂) act as nematocide, pesticide, hemolytic, 5-alpha reductase inhibitor;

2,6,10,18,22-Tetracosahexaene,2,6,10,19,23, hexamethyl (C₃₀H₅₀) used to treat Cancer prevention, anticancer activity, Cis cis,cis-7,10,13- Hexadecatrienal (C₁₆H₂₆O) having Antioxidant activity, Di-n-octylphthalate (C₂₄H₃₈O₄) having liver hitopathology enzyme activity, β-glucuronidase activity and 8, 11,14,Eicosatrienoic acid (Z,Z,Z) (C₂₀H₃₄O₂) has been reported to have anti-inflammatory, Insectifuge, Hypatoprotective, Antihistaminic, Antieczemic, Antiacen, 5-Alpha reductase inhibitor, Antiarthritic, Anticoronary activities. (Lalitha Rani *et al.*, 2009) [17].

Table 3. Chemical structure of the most prevailing compounds of the leaves extract of *Jasminum auriculatum* Vahl

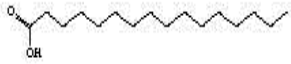

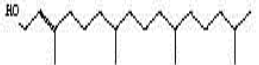
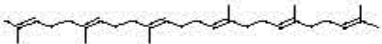

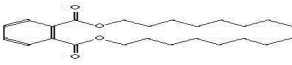
| Name of the compounds | Chemical structure of the compounds | Nature of the compounds | Biological activity |
|---|---|-------------------------|--|
| Ethanol leaf extract | | | |
| n-Hexadecanoic acid |  | Fatty acid | Nematicide, Pesticide, Lubricant, Antiandrogenic, Flavour, Hemolytic 5-Alpha reductase inhibitor |
| cis,cis,cis-7,10,13- Hexadecatrienal |  | Alcohol | Antioxidant activity |
| Phytol |  | Diterpene | Antimicrobial, Anti cancer, Anti-inflammatory Hypocholesterolemic, Nematicide, Anticoronary, Antiarthritic, Hepatoprotective, Anti -androgenic, |
| 2,6,10,14,18,22-Tetracosahexaene,2,6,10,19,23, hexamethyl |  | Triterpene | Cancer prevention, anticancer activity |
| 8, 11,14,Eicosatrienoic acid (Z,Z,Z) |  | Linolenic acid | anti-inflammatory, Insectifuge, Hypatoprotective, Antihistaminic, Antieczemic, Antiacen, Antiarthritic, Anticoronary |
| Di-n-octylphthalate |  | Plasticizer | liver histopathology enzyme activity, β -glucuronidase activity |

Table 4. FT-IR Spectral peak and functional groups obtained for the leaf extract of *Jasminum auriculatum* Vahl

| S.No | Peak value cm^{-1} | Types of stretching | Functional group | Corr. Area |
|------|-----------------------------|------------------------------|------------------------------------|------------|
| 1 | 524.64 | C-H wag (-CH ₂ X) | alkyl halides | 4.008 |
| 2 | 775.38 | C-H wag (-CH ₂ X) | alkyl halides | 0.455 |
| 3 | 893.04 | C-H stretch | aromatics | 0.024 |
| 4 | 1037.7 | C-N stretch | aliphatic amines | 19.109 |
| 5 | 1153.43 | C-H wag (-CH ₂ X) | alkyl halides | 0.68 |
| 6 | 1253.73 | C-H wag (-CH ₂ X) | alkyl halides | 1.649 |
| 7 | 1321.24 | C-N stretch | aromatic amines | 1.311 |
| 8 | 1386.82 | C-H rock | alkanes | 0.986 |
| 9 | 1409.96 | C-C stretch (in-ring) | aromatics | 1.69 |
| 10 | 1627.92 | N-H bend | primary amines | 33.656 |
| 11 | 1730.15 | C=O stretch | α,β -unsaturated esters | 0.359 |
| 12 | 1878.67 | C=C stretch | Anhydride | 0.002 |
| 13 | 1915.31 | C=C stretch | Alkene | 0.015 |
| 14 | 2166.06 | -C=C- stretch | alkynes | 0.343 |
| 15 | 2200.78 | -C≡C- stretch | alkynes | 0.001 |
| 16 | 2364.73 | C≡N stretch | nitriles | 0.48 |
| 17 | 2856.58 | C-H stretch | alkanes | 0.429 |
| 18 | 2924.09 | C-H stretch | alkanes | 5.549 |
| 19 | 3363.86 | O-H stretch | carboxylic acids | 134.832 |
| 20 | 3657.04 | O-H stretch, H-bonded | alcohol, phenol | 6.107 |

7. Conclusion

The investigation concluded that the stronger extraction capacity of ethanol could have been produced number of active constituents responsible for many biological activities. So that those might be utilized for the development of traditional medicines and further investigation needs to elute novel active compounds from the medicinal plants which may be created a new way to treat many incurable diseases.

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